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## **Answer For Basic Stoichiometry Pogil**

HS Chemistry POGIL Activity Page 2  
Basic Stoichiometry Model 2  $2A + 3B \rightarrow 5C + 4D$  If 3 mol A react, how many mol C are produced?  $3 \text{ mol A} \times 5 \text{ mol C} = 7.5 \text{ mol C}$   
2 mol A 5. If 21.0 mol A are reacted, how many mol C are produced? Using dimensional analysis, show how you calculated your answer. 6.

## **HS Chemistry POGIL Activity Basic Stoichiometry**

HS Chemistry POGIL Activity Topic: Stoichiometry. Basic Stoichiometry. Why? In this activity we will address the question: How do I convert between different chemical species in a given reaction? Model 1.  $2A + 3B \rightarrow 5C + 4D$ . 2 mol A produces 5 mol C 4 mol A

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produces 10 mol C. 3 mol B produces 4 mol D  
6 mol B produces 8 mol D

## **HS Chemistry POGIL Activity**

Chem 115 POGIL Worksheet - Week 4  
Moles & Stoichiometry Answers Key  
Questions & Exercises 1. The atomic weight of carbon is 12.0107 u, so a mole of carbon has a mass of 12.0107 g. Why doesn't a mole of carbon weigh 12 g? The atomic weight refers to the weighted average of masses of the isotopes comprising a naturally occurring sample of carbon.

## **Moles & Stoichiometry Answers Key Questions & Exercises**

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The Results for Pogil Stoichiometry Worksheet Answers. Structure Worksheet. Stoichiometry Worksheet 1 Answers

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Hs Chemistry Pogil Activity Basic Stoichiometry Answers Author: cdnx.truyeny.com-2020-11-16T00:00:00+00:01 Subject: Hs Chemistry Pogil Activity Basic Stoichiometry Answers Keywords: hs, chemistry, pogil, activity, basic, stoichiometry, answers Created Date: 11/16/2020 11:07:28 AM

## **Hs Chemistry Pogil Activity Basic Stoichiometry Answers**

Hs Chemistry Pogil Activity Basic Stoichiometry Answers 6 POGIL™ Activities for High School Chemistry 20. For each experiment in Model 2, determine the relationship between the

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independent and dependent variables, and write an algebraic expression for the relationship using variables that relate to those in the experiment (P internal, V, T or

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answer key Pogil the mole answer key. 00 u, so a mole of O<sub>2</sub> would have a mass of 32. 00 g and would contain 6. 022 x 10<sup>23</sup> O. 2 molecules. Each O<sub>2</sub> molecule is composed of two oxygen atoms, so one mole of O<sub>2</sub> contains two moles of oxygen atoms Pogil the mole answer

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HS Chemistry POGIL Activity Topic: Stoichiometry Basic Stoichiometry Why? In this activity we will address the question: How do I convert between different chemical species in a given reaction? Model 1  $2\text{A} + 3\text{B} \rightarrow 5\text{C} + 4\text{D}$  2 mol A produces 5 mol C 4 mol A

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produces 10 mol C 3 mol B produces 4 mol D 6 mol B produces 8 mol D 2 mol A reacts with 3 mol B 6 mol A reacts with 9 mol B 1.

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